



Direct numerical simulations of high speed reactive mixing layers

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Introduction

Problem description

- Direct numerical simulations: no turbulence models.
- High speed combustion regimes: Scramjet engines.
 - Gas mixtures: hydrogen, air ...
 - Multicomponent transport.
 - Detailed reaction mechanisms.
- Compressibility effects.
- Shocked configurations.

Introduction

Previous works

- Previous works dealt mainly with temporally evolving mixing layers, i.e. parallel flow assumption.
- Few examples of spatially developing mixing layers referred only to non reactive conditions (air flows).
- Infinitely fast chemistry assumptions/tabulated chemistry simplifications.

Governing equations

Fully compressible formulation, **multicomponent reactive mixtures**:

$$\partial_t(\rho) + \nabla \cdot (\rho \mathbf{u}) = 0, \quad (1)$$

$$\partial_t(\rho \mathbf{u}) + \nabla \cdot (\rho \mathbf{u} \otimes \mathbf{u} + P \mathbf{I}) = \nabla \cdot \boldsymbol{\tau}, \quad (2)$$

$$\partial_t(\rho \mathbf{e}_t) + \nabla \cdot [(\rho \mathbf{e}_t + P) \mathbf{u}] = \nabla \cdot (\boldsymbol{\tau} \mathbf{u} - \mathbf{q}), \quad (3)$$

$$\partial_t(\rho Y_\alpha) + \nabla \cdot (\rho Y_\alpha \mathbf{u}) = -\nabla \cdot (\rho Y_\alpha \mathbf{V}_\alpha) + \rho \dot{\omega}_\alpha, \quad \alpha \in \mathcal{S}. \quad (4)$$

$$\mathbf{e}_t = \mathbf{u} \cdot \mathbf{u} / 2 + \sum_{\alpha \in \mathcal{S}} h_\alpha Y_\alpha - \mathcal{R} T, \quad (5)$$

$$\boldsymbol{\tau} = \mu \left(\nabla \mathbf{u} + (\nabla \mathbf{u})^t \right) + (\kappa - 2\mu/3) (\nabla \cdot \mathbf{u}) \mathbf{I}, \quad (6)$$

$$\mathbf{q} = \sum_{\alpha \in \mathcal{S}} \rho Y_\alpha \mathbf{V}_\alpha (h_\alpha + \mathcal{R} T \tilde{\chi}_\alpha / \mathcal{W}_\alpha) - \lambda \nabla T, \quad (7)$$

$$\rho Y_\alpha \mathbf{V}_\alpha = -\rho Y_\alpha \sum_{\beta \in \mathcal{S}} D_{\alpha\beta} (d_\beta + X_\beta \tilde{\chi}_\beta \nabla T / T), \quad (8)$$

$$d_\alpha = \nabla X_\alpha + (X_\alpha - Y_\alpha) \nabla P / P. \quad (9)$$

Governing equations

$$P = \rho \mathcal{R} T / \mathcal{W}, \quad \mathcal{W} = (\sum_{\alpha \in \mathcal{S}} Y_{\alpha} / \mathcal{W}_{\alpha})^{-1}, \quad (10)$$

$$c_{p\alpha}(T) = \mathcal{R} \mathcal{W}_{\alpha}^{-1} \phi_{\alpha}, \quad (11)$$

$$h_{\alpha}(T) = \mathcal{R} T \mathcal{W}_{\alpha}^{-1} \varphi_{\alpha}. \quad (12)$$

Remarks

- ϕ_{α} and φ_{α} : polynomial representation (JANAF tables).
- Multicomponent transport (Soret and Dufour): EGLIB library^a.
- Chemical reactions: DVODE (CHEMKIN II library^b).

^aA. Ern and V. Giovangigli. Fast and accurate multicomponent transport property evaluation. J. Comput. Physics **120**, 105-116, (1995).

^bR. J. Kee, F. M. Rupley and E. Meeks. CHEMKIN-III: A Fortran Chemical Kinetics Package for the Analysis of Gas-Phase Chemical and Plasma Kinetics. Sandia National Laboratories (1996).

Numerical methods

Spatial discretization

- Convective fluxes: 7th order accurate Weighted Essentially Non Oscillatory (WENO) scheme^a.
- Molecular fluxes: 8th order accurate centered difference scheme.

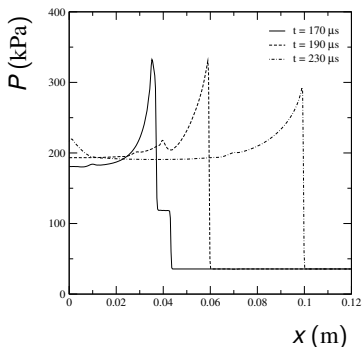
^aY. Shen and G. Zha. Improved seventh-order WENO scheme. AIAA Paper 2010-1451 (2010).

Temporal integration: Strang splitting technique^a

^aR. P. Fedkiw, B. Merriman and S. Osher. High accuracy numerical methods for thermally perfect gas flows with chemistry. J. Comput. Phys. **132**, 175–190, (1997).

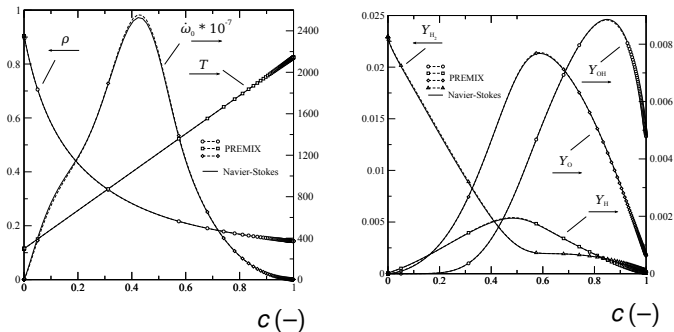
- 1 $t \rightarrow t + \Delta t/2$: DVODE solver.
- 2 $t \rightarrow t + \Delta t$: 3th order accurate Total Variation Diminishing (TVD) Runge Kutta scheme.
- 3 $t + \Delta t/2 \rightarrow t + \Delta t$: DVODE solver.

Multi-species reactive shock tube



- Riemann problem with a discontinuity at $t=0$.
- Focus on the competition between convection and reaction terms.
- Shock hits a solid wall boundary and reflects off.
- A reaction wave kicks in picking up steam and merges with the shock.

One-dimensional hydrogen/oxygen laminar premixed flame



- Similar orders of magnitude for convection, diffusion and reaction.
- PREMIX reference solution retained as the initial condition.
- Progress variable defined as: $c = (T - T_u)/(T_b - T_u)$.

Cheng approximate conditions

Overview

- Conditions representative of Scramjet engine operations.
- A reference benchmark for supersonic combustion modeling.
- Available experimental data^a.
- Different reaction mechanisms:
 - 1 Marinov reaction mechanism: 3 species, 1 step calibrated reaction.
 - 2 Jachimowski reaction mechanism: 13 species, 33 reactions.
 - 3 O' Conaire reaction mechanism: 9 species, 19 reactions.

^aT. S. Cheng, J. A. Wehrmeyer and R. W. Pitz. Raman measurement of mixing and finite-rate chemistry in a supersonic hydrogen-air diffusion flame. *Combustion and Flame* **99**, 157-173, (1994).

Shear layer configuration



$$L_x \times L_y = 350\delta_\omega^0 \times 90\delta_\omega^0 \text{ with } N_x \times N_y = 2085 \times 455.$$

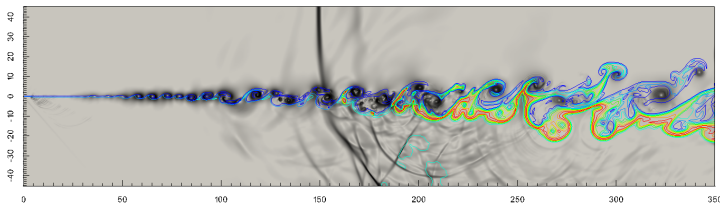
$$R_\omega = \bar{\rho} \Delta u \delta_\omega^0 / \bar{\mu} = 640, M_c = \Delta u / (c_f + c_o) = 0.4.$$

Fuel (top)	Oxidizer (bottom)
$u_f = 1949.08 \text{ m/s}$	$u_o = 954.55 \text{ m/s}$
$P_f = 109 \text{ KPa}$	$P_o = 109 \text{ KPa}$
$T_f = 545 \text{ K}$	$T_o = 1250 \text{ K}$
$Y_{\text{H}_2} = 1.0$	$Y_{\text{O}_2} = 0.245, Y_{\text{N}_2} = 0.58, Y_{\text{H}_2\text{O}} = 0.175$
$\rho_f = 0.049 \text{ Kg/m}^3$	$\rho_o = 0.28 \text{ Kg/m}^3$
$c_f = 1949 \text{ m/s}$	$c_o = 955 \text{ m/s}$
$M_f = 1.1$	$M_o = 1.34$

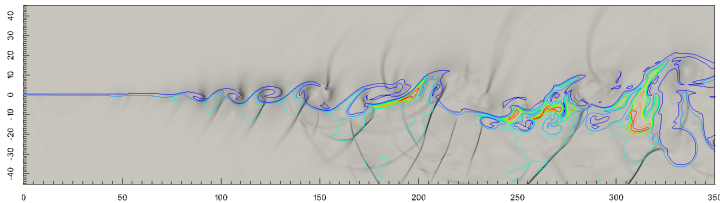
Instantaneous fields

Numerical Schlieren of the pressure field.

Temperature iso-contours: $T_{min} = 400$ K, $T_{max} = 3100$ K.



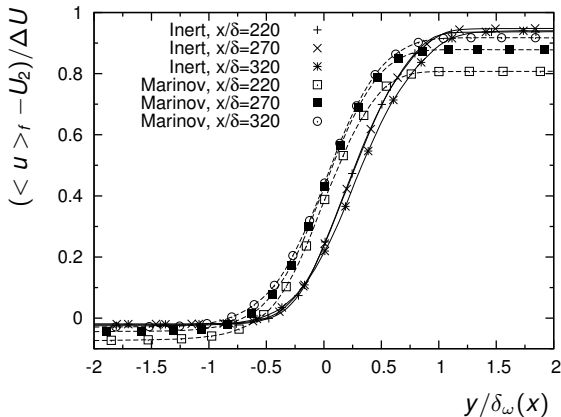
$M_c = 0.4$



$M_c = 0.8$

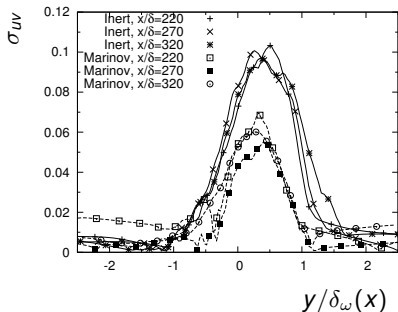
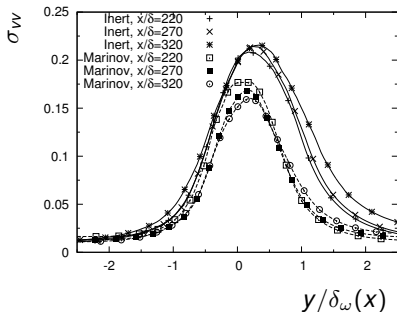
Self-similarity profiles ($M_c = 0.4$)

Favre averaged streamwise velocity component



Self-similarity profiles ($M_c = 0.4$)

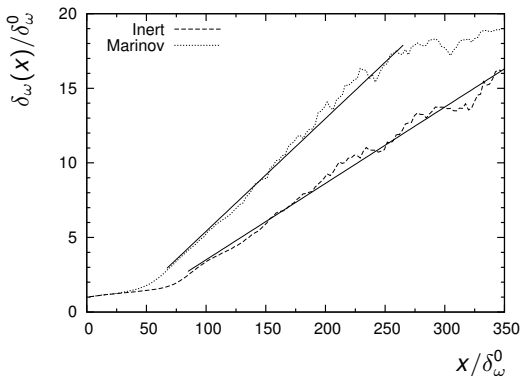
Reynolds shear stresses



$$\bullet \sigma_{wv} = \sqrt{\langle v'' v'' \rangle_f} / \Delta U.$$

$$\bullet \sigma_{uv} = \sqrt{\langle u'' v'' \rangle_f} / \Delta U.$$

Vorticity thickness growth rate ($M_c = 0.4$)



$$d\delta_\omega/dx = K_\delta(U_1 - U_2)/(U_1 + U_2), \quad K_\delta^I = 0.15, \quad K_\delta^R = 0.22$$

$$K_\delta^R/K_\delta^I \approx 1.47$$

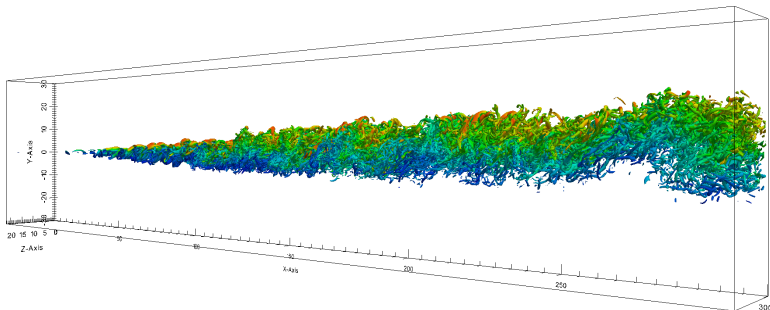
Cheng approx. conditions: conclusion

Re_θ	$\sigma_{uu_{max}}$	$\sigma_{vv_{max}}$	$\sigma_{uv_{max}}$	K_δ	Type	Reference
—	0.176	0.138	0.097	0.190	Experimental	Wynanski & Fiedler (1970)
—	0.190	0.120	0.114	0.160	Experimental	Spencer & Jones (1971)
450	0.180	0.140	0.100	0.163	DNS-3D	Bell & Mehta (1990)
800	0.160	0.130	0.100	0.130	DNS-3D	Rogers & Moser (1994)
90	0.200	0.290	0.150	0.143	DNS-2D	Stanley & Sarkar (1997)
160	0.220	0.220	0.110	0.150	DNS-2D	Inert simulation ($M_c = 0.4$)
160	0.220	0.180	0.070	0.220	DNS-2D	Reactive simulation ($M_c = 0.4$)

Concluding remarks

- Self-similarity in both inert and reactive cases.
- Shear stresses overestimated in 2D inert simulations.
- Heat release decreases significantly the Reynolds shear stresses and modifies K_δ .

3D simulations



- Iso-contour of $Q = 0.01 \times Q_{max}$ colored by the mixing variable, inert case at $M_c = 0.4$.
- $L_x \times L_y \times L_z = 300\delta_\omega^0 \times 60\delta_\omega^0 \times 22\delta_\omega^0$,
 $N_x \times N_y \times N_z = 1441 \times 307 \times 131$.

Conclusions and future works

Conclusions

- Cheng approximate conditions case confirms reduction of the Reynolds shear stresses.
- Moderate heat released effect with detailed reaction mechanisms compared to one step reactions.

Future works

- Influence of the inflow conditions (inflow oxidizer temperature, convective mach number. . .).
- 3D turbulent numerical simulations.
- Turbulent kinetic energy budget.
- Scalar turbulent transport.